



Organ-specific glucose uptake

Version: 1
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Summary: *(This area will include a brief description of what the protocol is used for and why someone would need to use it.)*

Glucose uptake in individual organs can be measured using a bolus injection of 2-deoxy-D-[1-¹⁴C] glucose, a non-metabolizable glucose analog, and by determining labeled metabolite levels in select tissues. Insulin resistance is characterized by reduced glucose metabolism and develops in obese mice.

Reagents and Materials: *(This should be a comprehensive list of stock solutions and material. The reagent list for the stock solutions is included in the reagent preparation area that is included at the end of this SOP.)*

Reagent/Material	Vendor	Stock Number
Poly-prep columns pre-filled with AG 1-X8 resin	Bio Rad	731-6211
0.2 M formic acid	Sigma	F0507
0.5 M ammonium acetate	Sigma	A1542

Protocol:

1. Survival surgery is performed to establish a chronic indwelling catheter at 5~6 days prior to experiment for intravenous infusion. (refer to M1023: Surgery-jugular vein cannulation)
2. Mice are fasted overnight (~15 hours) or for 5 hours prior to the start of experiment.
3. Place a mouse in a rat-size restrainer with its tail tape-tethered at one end.
4. Administer an intravenous bolus injection of 10 µCi of 2-deoxy-D-[1-¹⁴C] glucose (2-[¹⁴C]DG) in awake mice. Alternatively, intraperitoneal injection of 10 µCi of 2-[¹⁴C]DG may be used in awake mice.
5. After 30 min, rapidly freeze-clamp the tissues in liquid N₂, and store tissue samples in -80°C freezer for biochemical assay.
6. Biochemical assay is conducted using frozen tissue samples (e.g., skeletal muscle, adipose tissue, heart) to measure tissue levels of 2-[¹⁴C]DG-6-phosphate.
 - a) Prepare a heat block set to ~ 100°C.

- b) Prepare anion-exchange columns by washing with 5 ml of dH₂O.
 - c) Homogenize 50–100 mg of frozen tissue samples by adding ten times the volume of dH₂O (50 mg of tissue in 500 µl of dH₂O) in glass tubes using a tissue homogenizer.
 - d) Following homogenization, place the glass tubes in the heat block for 10 min, vortex for 2 sec, and then cool to room temperature.
 - e) Transfer the homogenized samples to microcentrifuge tubes using transfer pipettes and centrifuge at 16,000 × g for 5 min.
 - f) Add 33 µl of homogenate (supernatant) to 467 µl dH₂O in a scintillation vial labeled “total” sample.
 - g) Add 5 ml of scintillation cocktail, vortex, and count the samples for ¹⁴C using a liquid scintillation counter (total ¹⁴C samples).
 - h) Transfer 333 µl of homogenate (supernatant) to the anionexchange columns for the separation of 2-[¹⁴C]DG-6-P from 2-[¹⁴C] DG.
 - i) Wash the columns with 2 ml of dH₂O three times and collect the samples into a scintillation vial labeled “wash” sample.
 - j) Vortex the “wash” samples, and transfer 500 µl of “wash” samples to another set of scintillation vials to be counted for ¹⁴C using a liquid scintillation counter (wash samples containing 2-[¹⁴C] DG).
 - k) Elute the columns with 2 ml of 0.2 M formic acid/0.5 M ammonium acetate three times, and collect the samples into a scintillation vial labeled “eluate” sample.
 - l) Vortex the “eluate” samples, and transfer 500 µl of “eluate” samples to another set of scintillation vials to be counted for ¹⁴C using a liquid scintillation counter (eluate samples containing 2-[¹⁴C] DG-6-P).
7. The rate of glucose uptake in individual organs is determined using 2-[¹⁴C] DG. 2-[¹⁴C] DG is taken up by cells, phosphorylated by glucokinase to become 2-[¹⁴C] DG-6-P, and not further metabolized. Thus, organ-specific accumulation and level of 2-[¹⁴C] DG-6-P following a bolus injection of 2-[¹⁴C] DG reflect glucose uptake in individual organs.

Reagent Preparation: *(This area may have several different preparations with the table of contents below.)*

[Reagent 1](#)

[Reagent 2](#)

[Reagent 3](#)

Reagent 1: 0.2 M formic acid/0.5 M ammonium acetate

Reagents and Materials: formic acid, ammonium acetate, deionized water

Procedure:

1. Prepare 900 ml of dH₂O, and add 7.69 ml of formic acid.
2. Add 38.84 g of ammonium acetate, and adjust pH to 4.9±0.05 using dH₂O.
3. Add dH₂O to make a final solution volume of 1,000 ml.